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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/810,019	03/25/2004	Robert C. West	Q198-US1	2480
Quallion LLC 7590 02/02/2009				
P.O. Box 923127				
Sylmar, CA 91392-3127				
EXAMINER				
CRUPEAU, JONATHAN				
ART UNIT		PAPER NUMBER		
1795				
MAIL DATE		DELIVERY MODE		
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**Please find below and/or attached an Office communication concerning this application or proceeding.**

The time period for reply, if any, is set in the attached communication.

### Office Action Summary

**Application No.**

10/810,019

**Applicant(s)**

WEST ET AL.

**Examiner**

Jonathan Crepeau

**Art Unit**

1795

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --  
**Period for Reply**

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

**Status**

- 1) ☒ Responsive to communication(s) filed on 20 November 2008.
- 2a) ☒ This action is **FINAL**. 2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

**Disposition of Claims**

- 4) ☒ Claim(s) 1,3-5,9,10,12-14,16,17,19-24,26,27 and 55-62 is/are pending in the application.
- 4a) Of the above claim(s) \_\_\_\_\_ is/are withdrawn from consideration.
- 5) ☐ Claim(s) \_\_\_\_\_ is/are allowed.
- 6) ☒ Claim(s) 1,3-5,9,10,12-14,16,17,19-24,26,27 and 55-62 is/are rejected.
- 7) ☐ Claim(s) \_\_\_\_\_ is/are objected to.
- 8) ☐ Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement.

**Application Papers**

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on \_\_\_\_\_ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.  
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).  
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

**Priority under 35 U.S.C. § 119**

- 12) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☐ All b) ☐ Some \* c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
  2. ☐ Certified copies of the priority documents have been received in Application No. \_\_\_\_\_.
  3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

\* See the attached detailed Office action for a list of the certified copies not received.

**Attachment(s)**

- 1) ☐ Notice of References Cited (PTO-892)
- 2) ☐ Notice of Draftsperson's Patent Drawing Review (PTO-948)
- 3) ☐ Information Disclosure Statement(s) (PTO/SB/08)  
Paper No(s)/Mail Date \_\_\_\_\_
- 4) ☐ Interview Summary (PTO-413)  
Paper No(s)/Mail Date \_\_\_\_\_
- 5) ☐ Notice of Informal Patent Application
- 6) ☐ Other: \_\_\_\_\_

## **DETAILED ACTION**

### ***Response to Amendment***

1. This Office action addresses claims 1, 3-5, 9, 10, 12-14, 16, 17, 19-24, 26, 27, 55-61, and newly added claim 62. The claims are rejected under 35 USC 103 for substantially the reasons of record, and claims 1, 3-5, 9, 10, 12-14, 16, 17, 19-24, 26, 27, and 62 are newly rejected under 35 USC 112, second paragraph as necessitated by amendment. Accordingly, this action is made final.

### ***Claim Rejections - 35 USC § 112***

2. The following is a quotation of the second paragraph of 35 U.S.C. 112:

The specification shall conclude with one or more claims particularly pointing out and distinctly claiming the subject matter which the applicant regards as his invention.

3. Claims 1, 3-5, 9, 10, 12-14, 16, 17, 19-24, 26, 27, and 62 are rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention. Claim 1 has been amended to recite “terminal silicons and non-terminal silicons,” wherein “a portion of the silicons being linked to...”. The latter limitation is considered to be indefinite because it is not clear which silicons (i.e., the terminal silicons, non-terminal silicons, or both) must have the claimed linkage. Appropriate correction is required.

***Claim Rejections - 35 USC § 103***

4. Claims 1, 3-5, 9, 10, 12-14, 16, 17, 19-24, 26, 27, and 55-62 are rejected under 35 U.S.C. 103(a) as being unpatentable over Spiegel et al (U.S. Patent 6,447,952) in view of Lersch et al (U.S. Patent 5,606,077).

Spiegel et al. is directed to a chain-extended or crosslinked polysiloxane electrolyte for a lithium secondary battery (see abstract). The polysiloxane may comprise cyclic carbonate moieties on a terminal silicon thereof (see col. 4 line 25, col. 20, line 30). Regarding claims 9 and 62, each terminal silicon may comprise the carbonate moiety (see col. 3, line 40). Regarding claim 12, an oxygen may link the carbonate moiety to the terminal silicon (see col. 3, line 67). Regarding claim 21, as is apparent from the general formulas, the polysiloxane may have a molecular weight of less than 3000. Regarding claim 22, the ratio of alkali metal ions to cyclic carbonate groups is 1:30 to 1:5 (see col. 5, line 37). Since there are three "active oxygens" per carbonate group (see [0027] of instant specification), the ratio of [O]/[Li] is 15:1 to 90:1, which anticipates the claimed range. Regarding claim 24, the electrolyte is solid (see col. 5, line 52). Regarding claim 26, the polysiloxane is a member of an interpenetrating network. Regarding claim 27, the electrolyte has a conductivity of  $10^{-4}$  or  $10^{-2}$  S/cm or higher (see col. 3, line 15).

Spiegel et al. does not expressly teach terminal or middle silicons comprising a poly(alkylene oxide) moiety, as recited in claims 1, 10, 13, and 55.

Lersch et al. is directed to a polysiloxane having terminal silicons linked to cyclic carbonate moieties ("R2") and/or polyalkylene oxide moieties ("R5" / "R1") (see column 2; in particular the general formula and lines 20-29 and 60). Additionally, each silicon atom in the polymer backbone may have a cyclic carbonate and/or polyalkylene oxide moiety attached to it.

The moities may also contain an oxygen atom that is bonded to the silicon in the backbone (see col. 7, line 40).

Therefore, the invention as a whole would have been obvious to one of ordinary skill in the art at the time the invention was made because the substitution of one known element (the polymer of Lersch et al.) for another (the polymer of Spiegel et al.) would have yielded predictable results to one of ordinary skill in the art. It is submitted that the use of the polymer of Lersch et al. in the battery of Spiegel et al. would have provided the predictable result of a polysiloxane electrolyte having a high ionic conductivity. In column 5, Lersch et al. teach that their polymer has a high polarity, which is a result of the plurality of carbonate and polyalkylene oxide side chains containing active oxygen sites. Further, as disclosed in col. 3, line 6 of Spiegel et al., the carbonate moities allow the polymer of Spiegel et al. to have a high ionic conductivity. Thus, the skilled artisan can conclude from these disclosures that the increased use of polar moities having active oxygen sites would increase the polarity and ionic conductivity of a particular polysiloxane. Accordingly, this provides motivation to use the polysiloxane of Lersch et al. as the electrolyte of Spiegel et al.

It is further noted that the Lersch patent contains a citation to an article by Zhu et al. on the front of the patent. The Zhu et al. citation is specifically directed to the ionic conductivity of polysiloxanes containing carbonate side chains, which suggests the use of the polysiloxanes in a battery or other electrochemical device. This provides further evidence that the polymer of Lersch et al. would be suitable for use as an electrolyte in a battery such as that of Spiegel et al. Additionally, Lersch et al. teach that the polymers may be used for "electronic applications" in column 5, line 14.

Regarding claim 23, it would be obvious to use the polymer of Lersch et al. in either a liquid or solid form in the battery of Spiegel et al. The disclosure in column 5 of Lersch et al. indicates that the polymers are useable in liquids, and it would be obvious to employ them as part of a gelled system containing liquid in the battery of Spiegel et al.

Regarding claims 13, 14, 16, 17, 19, 20, and 55-61, which recite specific formulas with cross linkages, it is submitted that this subject matter would also be obvious to the skilled artisan. It is noted that the claimed formulas have a high degree of similarity to the general formula disclosed by Lersch et al., and the polymers of Lersch et al. may be routinely modified to result in polymers encompassed by the claimed formula. Furthermore, regarding the cross linkages (indicated as R4 in the instant claims), it would be obvious to perform cross-linking on the polymer of Lersch et al. Cross-linking is specifically disclosed in the Spiegel et al. reference and is known to increase strength and molecular weight of a material, the former being particularly valuable in a solid electrolyte. Accordingly, the artisan would be motivated to cross-link the polymer of Lersch et al.

#### ***Response to Arguments***

5. Applicant's arguments filed November 20, 2008 have been fully considered but they are not persuasive. Regarding the citation of the Zhu et al. article on the front of the Lersch patent, Applicants state that "it is critical to know whether the Examiner is relying simply on the title of this article or is incorporating the contents of this article into the rejection." In response, only the title of the article is being relied on by the Office. The position is maintained that a skilled

artisan would understand that the disclosure of “ionic conductivity” in the title of the article suggests an end application involving ion exchange, in particular, an electrochemical device. Furthermore, as stated in the rejection above, the presence of this citation is only one of several factors that lead to a conclusion of obviousness. Throughout the Lersch and Spiegel patents, importance is placed on ionic conductivity and polarity, and the stated motivation for making the combination of references is to provide a polymer having high ionic conductivity as the electrolyte of Spiegel. Further, the position is maintained that the use of the polymer of Lersch in the battery of Spiegel would provide a predictable result.

### ***Conclusion***

6. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, THIS ACTION IS MADE FINAL. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event,

however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Jonathan Crepeau whose telephone number is (571) 272-1299. The examiner can normally be reached Monday-Friday from 9:30 AM - 6:00 PM EST.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Patrick Ryan, can be reached at (571) 272-1292. The phone number for the organization where this application or proceeding is assigned is (571) 272-1700. Documents may be faxed to the central fax server at (571) 273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

/Jonathan Crepeau/  
Primary Examiner, Art Unit 1795  
February 2, 2009